

## Factors affecting the localization of transformations of nitrogen and methane in soils

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### ABSTRACT

Biological, chemical, and physical factors regulate the growth and activity of bacteria responsible for nitrification, denitrification, and methane oxidation in soils. The restriction of activities to certain soil horizons or to certain microenvironments can affect important environmental processes including losses of fertilizer nitrogen, and the production and consumption of the atmospheric trace gases nitric oxide, nitrous oxide and methane.

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### Introduction

Among the many factors likely to govern the growth and activities of microorganisms involved in transformations of nitrogen and methane, the most important are the availability of substrates (organic or inorganic reductants) and electron acceptors (principally O<sub>2</sub>), the presence of potential inhibitory compounds, and the ambient pH of the environment. Here we consider these in relation to N transformations, mainly nitrification and denitrification, and CH<sub>4</sub> oxidation.

### Substrate effects

Nitrification is limited by the availability of NH<sub>4</sub><sup>+</sup> in some soils, but CO<sub>2</sub>, the source of carbon is rarely limiting. Denitrification was recognized long ago to be frequently limited by available carbon, but it was Parkin [18] who demonstrated the importance of active "hot spots", located in or around fragments of organic matter, in causing the high spatial variability often found in soil assays of denitrification. Later microsensor studies by Højberg et al. [12] determined both N<sub>2</sub>O and O<sub>2</sub> in large water-saturated aggregates on the surface of which a disc of clover leaf was pressed. Incubations in the presence and absence of C<sub>2</sub>H<sub>2</sub> demonstrated clearly that in the region below the disc, leaf components stimulated both O<sub>2</sub> consumption and denitrification to levels that were much greater than were observed in zones well outside the leaf disc.

Methanotrophic bacteria responsible for oxidation of CH<sub>4</sub> are widely distributed in nature. They are especially active adjacent to methanogenic habitats, such as in the surface layers of peats and in landfill cover soils. In these environments, the bacteria are exposed to relatively high CH<sub>4</sub> concentrations and thus the so-called low affinity (high K<sub>m</sub>) methanotrophs seem dominant throughout the aerobic zone. On the other hand, many aerobic soils exhibit CH<sub>4</sub>-oxidizing activity and consume atmospheric CH<sub>4</sub>, but despite the fact that the source is the atmosphere, this activity is frequently (especially in forest soils) restricted to a subsurface layer. These bacteria are of the high affinity (low K<sub>m</sub>) type, consuming CH<sub>4</sub> at concentrations two or three orders of magnitude lower than exist in the low affinity systems [11]. It is not clear how such methanotrophs can show maintenance and

### *Microbial Biosystems: New Frontiers*

*Proceedings of the 8<sup>th</sup> International Symposium on Microbial Ecology*

*Bell CR, Brylinsky M, Johnson-Green P (ed)*

*Atlantic Canada Society for Microbial Ecology, Halifax, Canada, 1999*

growth at such low CH<sub>4</sub> concentrations unless some other C<sub>1</sub> substrate, such as methanol, is at least transiently available. Possible reasons for their restriction to the subsurface layer are discussed later.

### Inhibitor effects

Polyphenols and tannins have been suggested as inhibitors responsible for the little or no nitrification occurring in some soils [17]. Monoterpenes are released from some forest systems at rates up to 5 mg m<sup>-2</sup> h<sup>-1</sup> [13] and have been implicated, by Rice and White and their coworkers, in affecting nitrification in forest soils [21,24]. However, the topic is controversial, and Bremner and McCarty [7] claimed that the apparent inhibition seen in earlier studies was due to immobilization of NH<sub>4</sub><sup>+</sup> and/or NO<sub>3</sub><sup>-</sup> by microbial activity stimulated by the organic carbon added in the form of the monoterpene. More recent work by Ward et al. [23] shows conclusively that several monoterpenes (especially  $\alpha$ -pinene and R+-limonene) inhibit oxidation of NH<sub>4</sub><sup>+</sup> by *Nitrosomonas europaea* in pure culture under NH<sub>4</sub><sup>+</sup>-sufficient conditions (Table 1). Both immobilization and inhibition may occur in some systems.

**Table 1.** Effect of monoterpenes on the oxidation of NH<sub>4</sub><sup>+</sup> by *Nitrosomonas europaea* (from [23]) and of CH<sub>4</sub> by *Methylosinus trichosporium* and some forest soil samples (from [3,5])

Addition	NH <sub>4</sub> <sup>+</sup> oxidation	CH <sub>4</sub> oxidation			
	<i>Nitrosomonas europaea</i>	<i>Methylosinus trichosporium</i>	Mixed 7-12 cm	Pine 5-10 cm	Aspen 5-10 cm
	10 $\mu$ g/mL	245 $\mu$ g/mL	10 mg/g	10 mg/g	5 mg/g
None	100	100	100	100	100
$\alpha$ -pinene	11	4	0	4	50
R+-limonene	35	16		39	35
$\alpha$ -terpinene		75	0	30	34
$\beta$ -myrcene	45	93	16	38	85
$\alpha$ -pinene oxide		103	235		
$\pm$ -linalool		32	116		
$\alpha$ -terpineol		52	136		

Mixed denotes a mixed hardwood stand

We have speculated that the lack of CH<sub>4</sub> oxidation in the surface layer of many forest soils may be caused by a similar effect, and, indeed, extracts of such layers do inhibit CH<sub>4</sub> oxidation in both soils and pure cultures [2]. We have also shown that CH<sub>4</sub> oxidation in some forest soil samples is inhibited significantly by  $\alpha$ -pinene, R+-limonene,  $\alpha$ -terpinene, and  $\beta$ -myrcene [3], and that some of these compounds also inhibit CH<sub>4</sub> oxidation by *Methylosinus trichosporium* in pure culture [5] (Table 1). The concentrations used in these studies were fairly ecologically relevant. However, conclusive proof that this effect is responsible for the observed localization of nitrification or CH<sub>4</sub> oxidation in a particular natural system remains difficult to obtain.

### pH effects

The autotrophic nitrifiers are notoriously acid-sensitive, and, in pure culture, both they and the known methanotrophs have pH optima near neutrality. Despite these facts, both groups are reported to be active in acidic forest soils. In such reports, it is usually assumed that nitrification activity that is sensitive to C<sub>2</sub>H<sub>2</sub> can be attributed to autotrophic rather than heterotrophic nitrifiers. However, acid tolerant NH<sub>4</sub><sup>+</sup> oxidizers have not been isolated from such soils, and it is often speculated that more alkaline microsites in the soil are involved. More recently, de Boer and coworkers [9] pointed out the possible significance of aggregations of cells of *Nitrosospora* spp. surrounded by cells of *Nitrobacter* in a nitrifying Douglas fir soil with a pH of 3.7. Thus it appears that this may be an example of a common phenomenon in which colonial growth of microorganisms protects them from unfavourable environmental conditions.

Methanotrophic activity that occurs in acidic peats and forest soils may be subject to microsite protection of some kind, it may be due to colonial aggregations not so far reported, or it may involve different methanotrophs having different pH optima. We showed that methanotrophs active in peats were not well adapted to the ambient pH and slurries showed optimum pH values that were 0.5 to 1.0 unit higher than the *in situ* pH [10]. We studied this aspect also in forest soils by measuring atmospheric CH<sub>4</sub> uptake rates at a range of pH values both in soil slurries and in bacteria extracted from the soil by Nycodenz density gradient centrifugation [19] and then washed. We found that the pH optima of slurries were generally rather close to the ambient soil pH, but that extracted bacteria showed pH optima for CH<sub>4</sub> uptake that were higher, in the range 5.8 to 6.0, but still about one pH unit lower than the pH optima for two known methanotrophs studied, *Methylobacter luteus* and *Methylosinus trichosporium* [6]. Indeed, for the extracted bacteria, the decline in activity on each side of the optimum was very steep, suggesting that they are very sensitive to even small changes in pH, and that they would be unlikely to grow well in the culture media normally used for the known methanotrophs.

It also appeared in our study that the indigenous methanotrophs in the acidic soils were roughly 50% resistant to the high speed blending used before extraction, but that methanotrophs in neutral agricultural soils were completely inactivated by blending, and no activity could be extracted. However, in the acidic previously-inactive surface horizon of a mixed hardwood soil enriched by exposure to 5% CH<sub>4</sub> the newly active methanotrophs were completely resistant to blending, and, furthermore, about 50% of the activity could be extracted [6]. These observations, all taken together, suggest that the atmospheric CH<sub>4</sub> consumers active in acidic forest soils are different from, or in a physiological state different

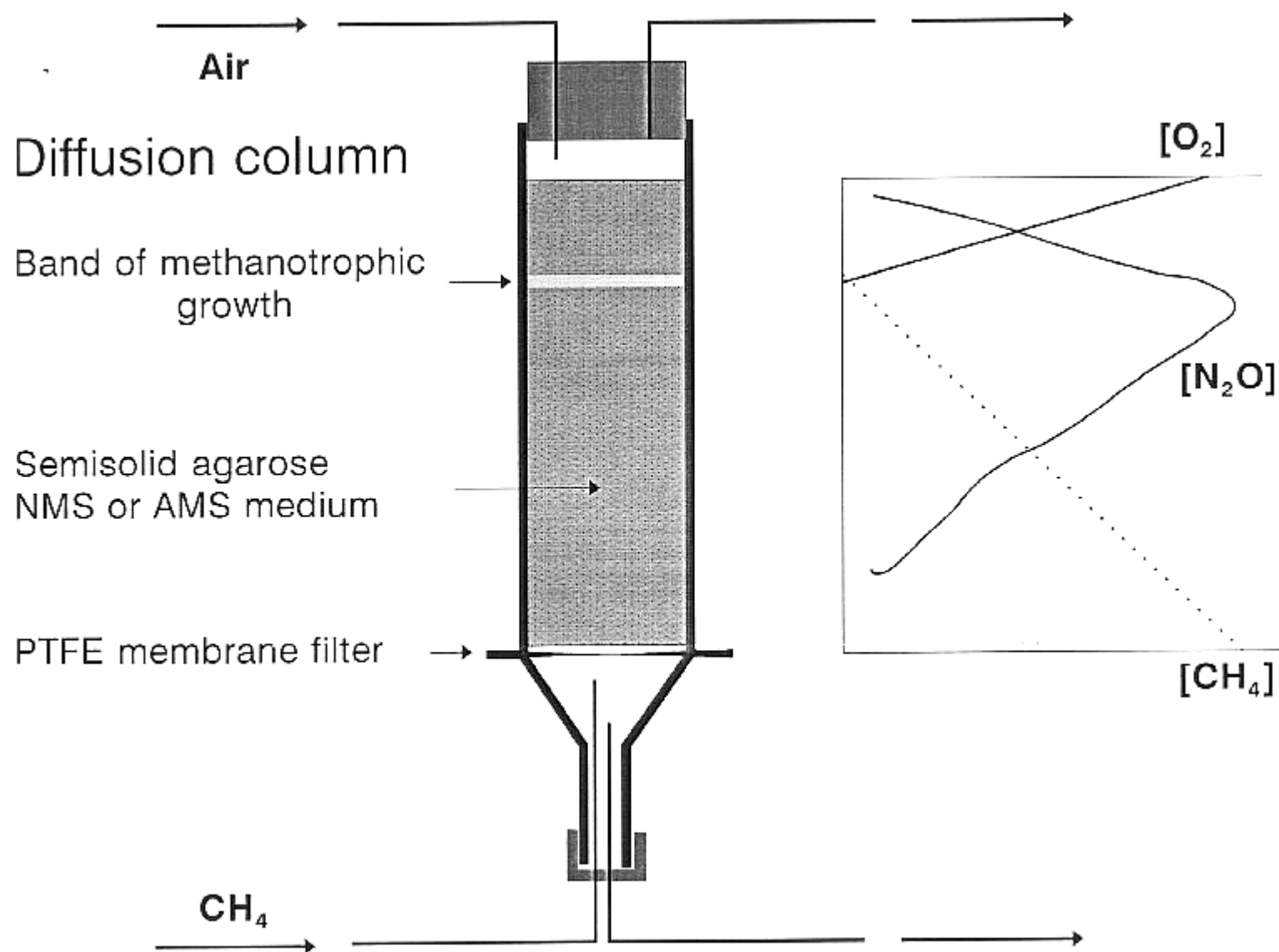
from, those that are enriched by exposure to high CH<sub>4</sub> concentrations and from the known methanotrophs of laboratory culture.

## **O<sub>2</sub> effects**

Denitrifying reductase genes are repressed by O<sub>2</sub> and their activity is generally inhibited by O<sub>2</sub>, so denitrification is mostly limited to anaerobic environments. Recent reports of aerobic denitrification in cultures [15] have left open the question of the ecological significance of this phenomenon. However, the microsensor studies of Revsbech and coworkers have made it possible to determine microprofiles of O<sub>2</sub> and denitrification substrates and products in a range of soils, sediments, and biofilms. They reported that at no time was there N<sub>2</sub>O evidence that denitrification occurred in the aerobic zone at > 15 μM O<sub>2</sub> [8].

The microscale distribution of methanotrophs and their activity in soil is difficult to study. Stratified lakes, however, often show a thin plate of activity 0.25 to 1.0 m deep in the metalimnion and close to the aerobic-anaerobic interface. It is suggested that nitrogen limitation could require microaerophilic N<sub>2</sub>-fixing growth, but our laboratory studies with N-sufficient cultures of three methanotrophs showed no evidence of microaerophilism [20]. Furthermore, our studies with diffusion-controlled growth in semi-solid agarose columns (Fig. 1) reproduce the type of stratification observed in lakes. Inoculation of an entire column with soil suspension results in the establishment of a band of methanotroph biomass and activity where the tails of the linear gradients of O<sub>2</sub> and CH<sub>4</sub> meet, whether the mineral salts medium contains sufficient NO<sub>3</sub><sup>-</sup> (NMS) or sufficient NH<sub>4</sub><sup>+</sup> (AMS) as N source. Based on the concentration gradients, the supply rates (and therefore consumption rates) are generally in the ratio of 1.5-1.9:1 O<sub>2</sub>:CH<sub>4</sub> in agreement with other pure culture studies [1]. Simulation of growth in such columns, using the programme of Koch [14], but inserting methanotroph parameters from the literature, reproduces the position of the growth band and the concentration gradients of O<sub>2</sub> and CH<sub>4</sub> seen in the experimental column. It is noteworthy that nitrogen transformations are often associated with the growth of methanotrophs. For example, in NMS columns, N<sub>2</sub>O is produced by associated denitrifiers, and in AMS columns, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> may be produced, either by methanotrophic nitrification or by associated autotrophic nitrifiers during the period of O<sub>2</sub> depletion in the column [1,4].

A perplexing phenomenon in relation to O<sub>2</sub> is the frequent occurrence of obligate aerobes in anaerobic environments. For example, potential activities and counts of both nitrifiers and methanotrophs occur in a lake sediment for several centimetres below the one mm or so O<sub>2</sub> penetration depth [22]. They could be transported by bioturbation and are known to survive periods of anaerobiosis, but maintenance and growth for long periods under anaerobic conditions is so far not yet explained. This is but one example of questions that are still open in relation to these interesting organisms.



**Fig. 1.** A gel-stabilized column for studying the growth of methanotrophs in counter-gradients of O<sub>2</sub> and CH<sub>4</sub> (For details, see refs. [1] and [4]).

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